## CHM B20H3 and B23H3

## Chemical Thermodynamics and Elementary Kinetics 2019-2020

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**Textbook:** *Physical Chemistry* 11th Ed., vol. 1

by Atkins, de Paula and Keeler

<b>Grading:</b>	CHM B20		<b>CHM B23</b>	
2 in-class tests	(15% + 25%)	= 40%	(10%+20%)	= 30%
5 assignments	5 x 5%	= 25%	5 x 3%	= 15%
Laboratory & tutorial				20%
Final exam		35%		35%

Office Hours: Monday & Wednesday 10-12 or by appointment

If you miss a test or an assignment (without my prior written consent), you will need to provide an official UTSC medical note (or do some really good talking!) in order to avoid a mark of zero.

Except under special circumstances, arranged with me, there will be no make-up tests. If you miss a test, the weighting will be adjusted to give more to the other test and to the final exam.

Cheating on tests and assignments will be dealt with very harshly. Although I do encourage group discussion of how to solve problems, I expect each student to produce an individual assignment to hand in. It is very obvious to the markers when one person has done the work and others have copied it.

## Lecture Schedule (Note that this may change somewhat)

Week	a Dates	Chapter reference	notes			
1	Sept 4	Chapter 1A				
2	Sept 9, 11	Chapter 1B,C				
3	Sept 16, 18	Chapter 2A,B,C	Assignment 1 due Sept 18			
4	Sept 23, 25	Chapter 2D,E; 3A,B,C				
5	Sept 30, Oct 2	Chapter 3D,E; 4A,B	Test 1 Oct 2 (Chapter 1-3)			
6	Oct 7, 9	Chapter 5A,B,C	Assignment 2 due Sept 25			
READING WEEK						
7	Oct 21, 23	Chapter 5F; 6A,B,C,D				
8	Oct. 28, 30	Chapter 16	Assignment 3 due Oct 30			
9	Nov 4, 6	Chapter 17A,B,C,E	Test 2 Nov 6 (Chapter 4,5,6,16)			
10	Nov 11, 13	Chapter 17F,D; 18A,B	Assignment 4 due Nov 13			
11	Nov 18, 20	Chapter 18C; 19A				
12	Nov 25, 27	Chapter 19B,C	Assignment 5 due Nov 27			
13	Dec 2	Review				

I will try to reserve the Wednesday lectures for problem-solving, discussions and for going over some new mathematical concepts

## Lab and Tutorial Schedule

Wee	k Dates	<u>Experiment</u>	P01	P02
1	Sept 4			
2	Sept 12		T1	
3	Sept 19	Thermochemistry	L1	<b>T1</b>
4	Sept 26		<b>T2</b>	L1
5	Oct 3		L2	<b>T2</b>
6	Oct 10	Equilibrium constant and sun anthal	T3	L2
7	Oct 24	Equilibrium constant and rxn enthal	L3	<b>T3</b>
8	Oct. 31	Vinatios of fluorescence quenching	<b>T</b> 4	L3
9	Nov 7	Kinetics of fluorescence quenching	L4	114
10	Nov 14	Adsorption from solution	T5	<b>L4</b>
11	Nov 21	Adsorption from solution	L5	<b>T5</b>
12	Nov 28			<b>L5</b>